The University of Tokyo, Komaba Graduate School of Arts and Sciences, College of Arts and Sciences

## Expected background knowledge in Chemistry for students on the PEAK Environmental Sciences course

MPEAK Programs in English at Komaba

X - no coverage	L - very limited coverage	Р-р	artial cover	age	✓ - cc	overed	
Su	bject	Coverage in common s		chool examination s		vstems	
	,	GCSE	IB Standard	SAT	A-level	IB Higher	AP
1. Chemical nature of substar	nces - Inorganic substances				•		
a) The elements Nature of typical elements and the periodic table 1st to 3rd period and K, Ca		Р	P	~	~	~	~
b) Compounds Compounds of typical elements Compounds of first row transition elements Oxides and Hydroxides		Ρ	~	~	~	~	~
c) Detection of ions Reactions of typical ions - colour reactions and precipitation reactions including Ag <sup>+</sup> , Cu <sup>2+</sup> , Zn <sup>2+</sup> , Ba <sup>2+</sup> , Ca <sup>2+</sup> , Sr <sup>2+</sup> , Mg <sup>2+</sup> , Na <sup>+</sup> , K +, Al <sup>3+</sup> , Pb <sup>2+</sup> , Fe <sup>2+</sup> , Fe <sup>3+</sup> , Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup> , HCO <sub>3</sub> <sup>-</sup> , l <sup>-</sup>		Р	~	~	~	~	•
2. Chemical nature of substar	nces - Organic substances		•		•	•	
a) Chemical features of organic compounds Aliphatic and aromatic compounds Determination of molecular formula Structures of organic compounds and their classification Structural formula and isomers		L		Ρ	<b>`</b>	~	~
b) Compounds containing carbon, hydrogen, and oxygen Hydrocarbons, Alcohols, Ethers, Aldehydes, Ketones Carbonic acids, Phenols Fats/oils and detergents		×	Р	L	~	~	Р
	nces - Reactions of organic comp	ounds					
Substitution, addition and elimination reactions		L	×	L	~	<b>~</b>	L
4. States of matter					-		
a) Three states of matter Melting and evaporation, meltin The ideal gas law	g point and boiling point	Ρ		~	~		~
b) Mixtures Partial pressure of a gas Solutions and solubility Nature of dilute solutions and colligative properties		×	Р	~	~	Р	~
5. Chemical reactions							
a) Qualitative treatment of react	ion rates	Р	<b>v</b>	~	~	~	~
<ul> <li>Thermodynamics - heat of re Fhermochemical equations</li> </ul>	action	Р	~	~	~	~	~
c) Chemical equilibrium Reversible reactions and Le Châtelier's Principle		Р	~	~	~	~	~
d) Acid base reactions, neutralizations Hydrogen ion concentration and pH		Р	~	~	~	~	~
e) Redox reactions Oxidation and reduction Electrolysis onization tendency Electrical cells (batteries)		×		~	~		~

Subject	Cove	Coverage in common school examination systems					
	GCSE	IB Standard	SAT	A-level	IB Higher	AP	
6. Structure of matter							
a) Atomic structure	P	P	~	~	<b>v</b>	~	
Models of atomic structure							
Shell structure of electrons							
Atomic structure and the periodic table							
b) Chemical bonding	P	<b>~</b>	Р	~	<b>~</b>	~	
lonic bonding							
Covalent bonding							
The structure and nature of matter							

🗙 - no coverage 🛛 L - very limi	ted coverage P - partial cove	erage 🖌 - covered
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